



# Soil Mineral Management - Understanding Your Soil Test

By Graeme Sait

# Getting Back To Root Causes

- Poor quality crops and increasing pest pressure are never an **accident** or a question of bad luck.
- A fungal disease is not a **deficiency of a fungicide**.
- Medicine, veterinary science and agricultural science have all become based upon the **treatment of symptoms**.
- The problem is not solved. The chemicals serve as a temporary **bandaid**.
- Nutrition Farming is all about getting back to the **root cause** and solving the problem.



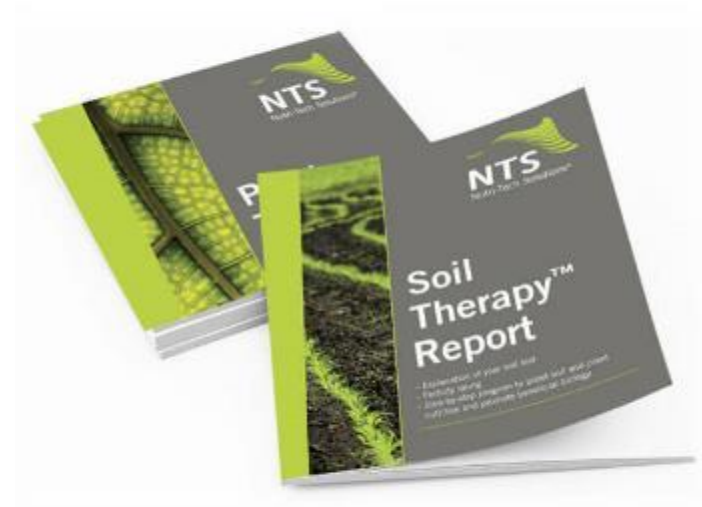
## Start With a Good Soil Test

**Crop nutrition** should never involve guesswork. Informed nutrition is essential to avoid driving blind.

There is a **specific balance** between soil minerals that determines productivity.

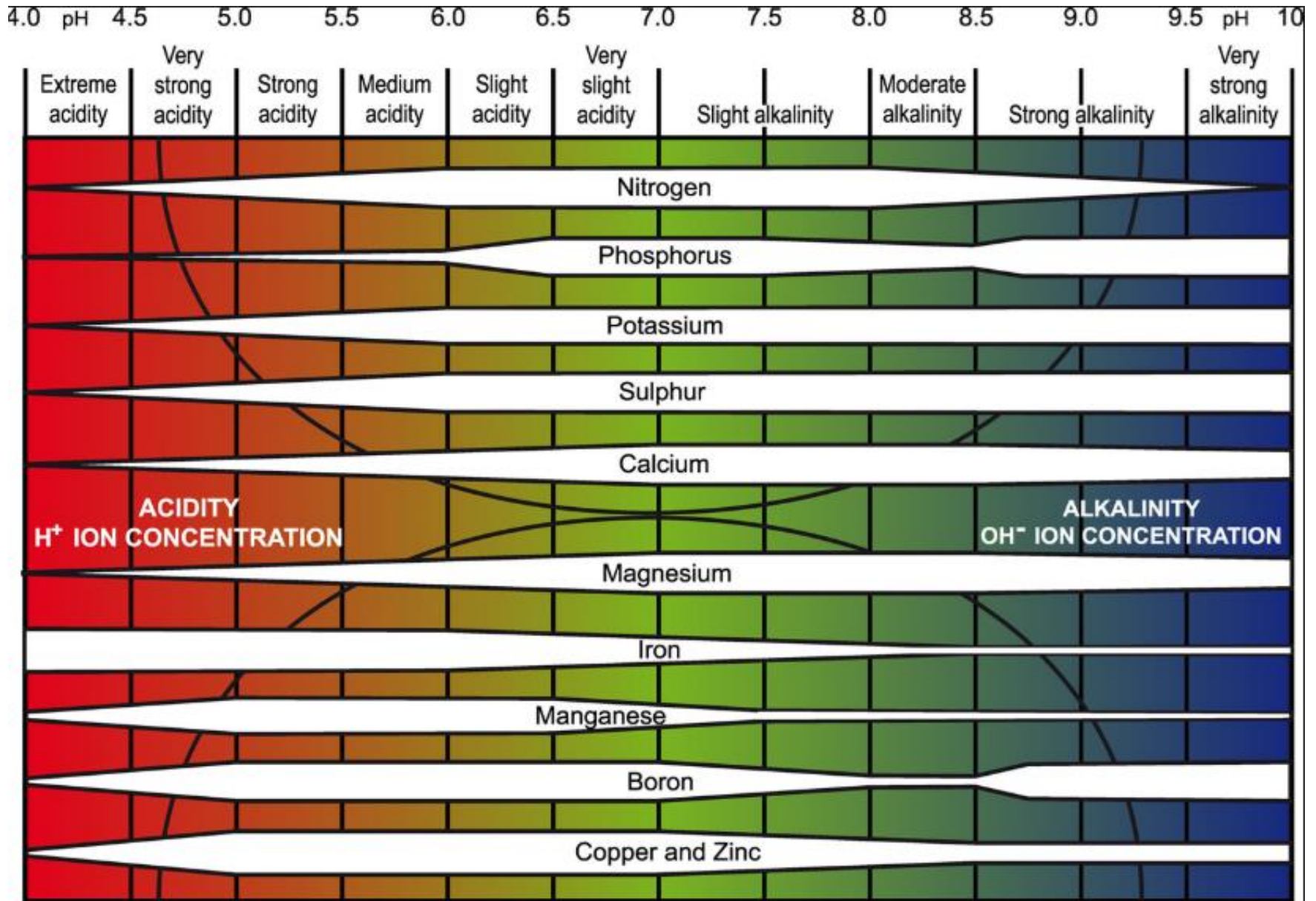
Seek a lab or consultant (NTS Soil Therapy™, Brookside, etc) with an understanding of how to achieve this productive state of **balance**.

If necessary corrections are unaffordable then liquids can be used for **crop nutrition** (liquid inject and foliar).



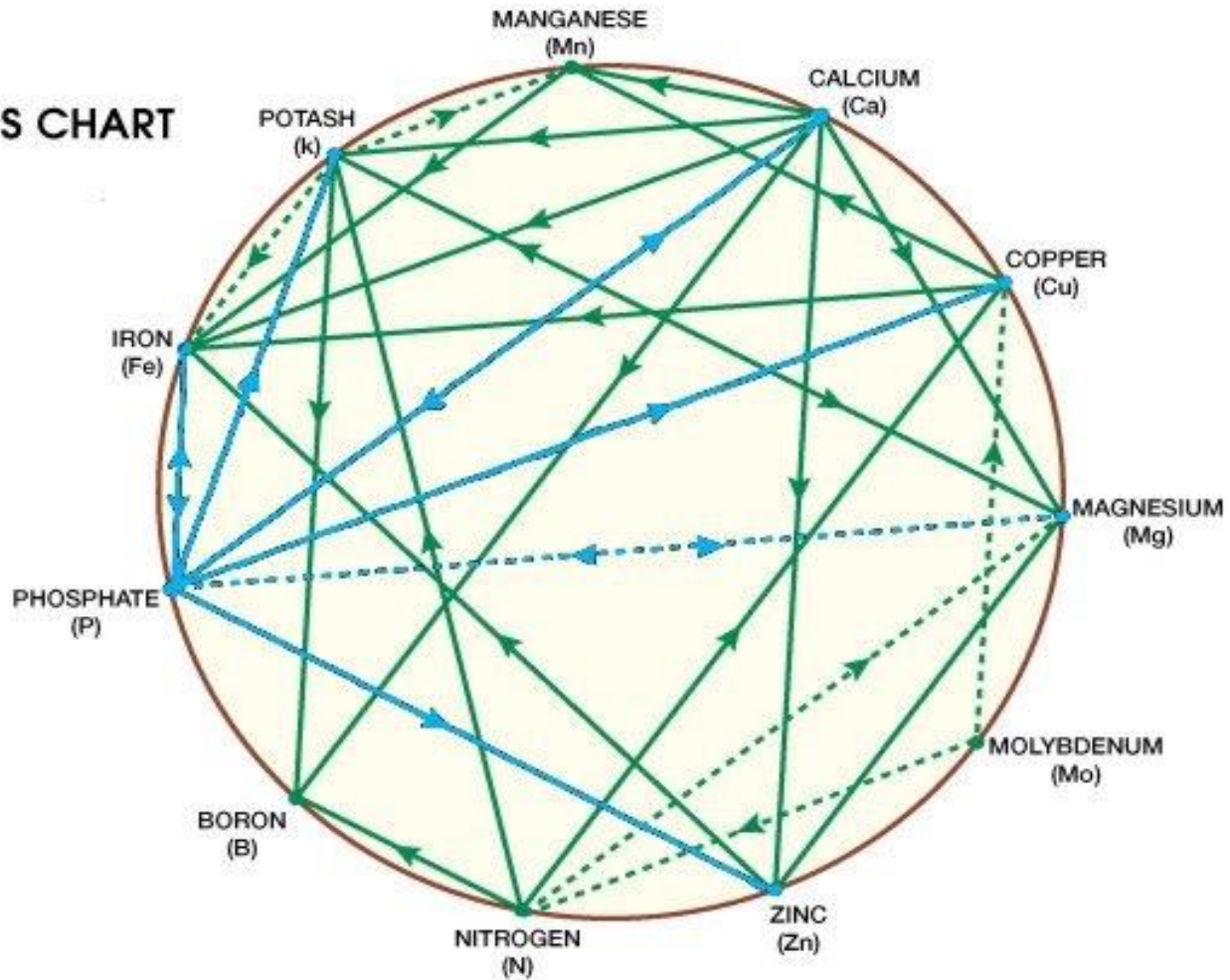


# Cutting-Edge Nutrition Strategies for Coffee



# Cutting-Edge Nutrition Strategies for Coffee

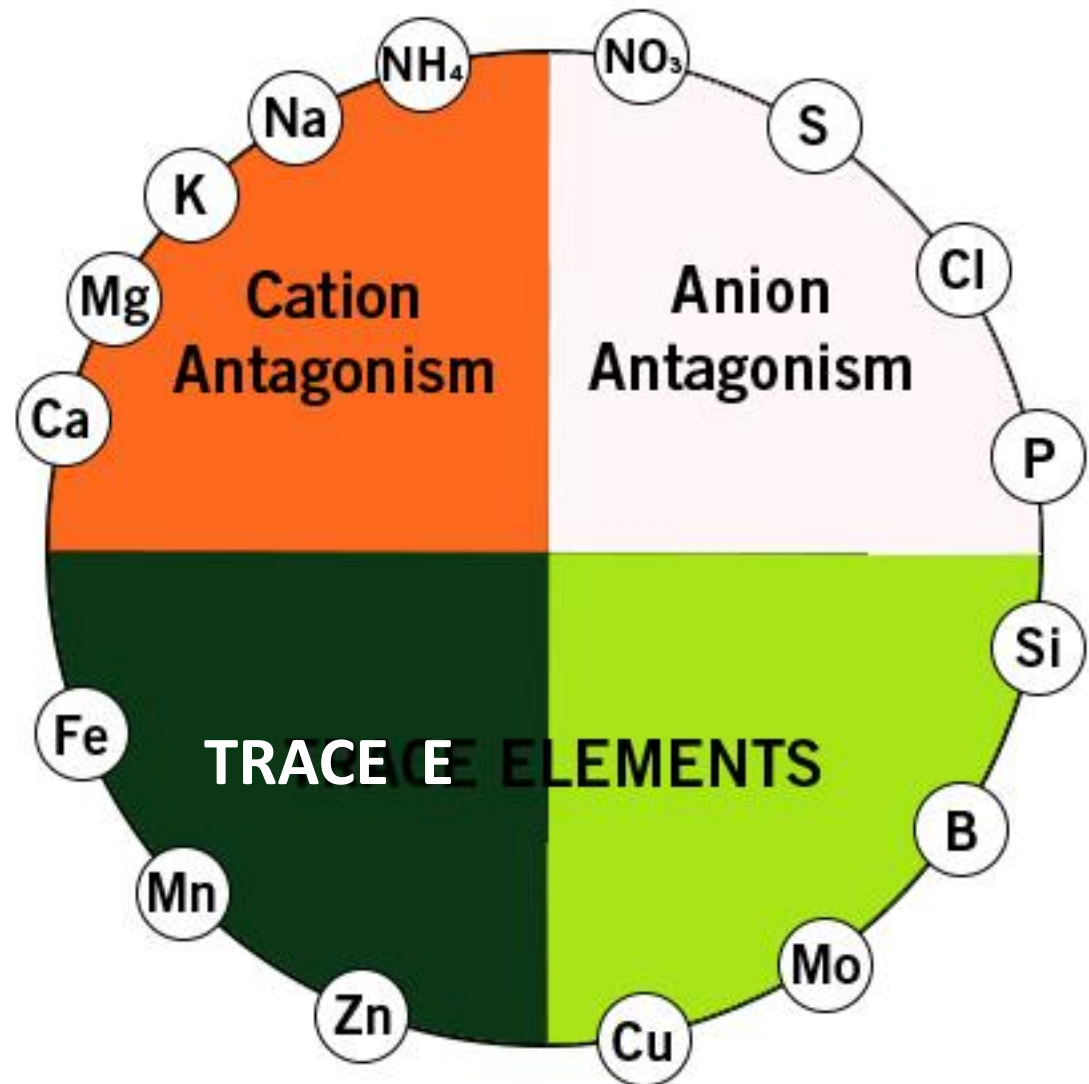
## MULDER'S CHART



- Antagonism** ——— A decrease in availability to the plant of a nutrient by the action of another nutrient. (see direction of arrow)
- Stimulation** - - - - - An increase in the need for a nutrient by the plant because of the increase in the level of another nutrient.

# Cutting-Edge Nutrition Strategies for Coffee

- This pie chart highlights how the minerals **impact** one another.
- Too much of any major cation or anion will **shut down** another mineral in the same group.
- Everything within each quartile **directly impacts** everything else within that segment of the pie i.e. too much gypsum effects phosphorus, chloride, and nitrate nitrogen uptake.





# Soil Mineral Management

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## Soil Test Understanding

- It is critically important to **reclaim responsibility** for your own enterprise, rather than dependence upon those with vested interests.
- Ideally, you need to fully understand the **lessons** from your soil test.
- You will then be equipped to develop your own **nutrition program**.
- **Knowledge** is your empowerment, in the regenerative approach.



# Soil Tests – 5 Productive Add-Ons

1. **Chloride** – often not required but essential when needed (for vegetative growth).
2. **Silicon** – as this mineral is missing in most soils and it determines diseases resistance.
3. **Molybdenum** – Needed to access free nitrogen from the air.
4. **Cobalt** – supports nitrogen-fixing organisms.
5. **Selenium** – so important for both animal and human health (and plant health)





# Soil Mineral Management

## The Key Ratios

1. Ca : Mg ratio
2. Mg : K ratio
3. P : S ratio
4. P : Zn ratio
5. K : Na ratio
6. Fe : Mn ratio

ALBRECHT CATEGORY	YOUR LEVEL	IDEAL LEVEL	NUTRIENT STATUS		
			LOW	MEDIUM	HIGH
CEC	31.45				
TEC	34.56				
Paramagnetism	350	200 +			
pH-level (1:5 water)	6.40	6.3			
Organic Matter (IR Gas Anal.)	6.52 %	4 - 10 %			
Conductivity (1:2 water)	0.20 mS/cm	0.2 - 0.6 mS/cm			
Ca / Mg Ratio	1.95 :1	7.00 :1			
Nitrate-N (Morgan)	29.7 ppm	10 - 20 ppm			
Ammonium-N (Morgan)	11.2 ppm	10 - 20 ppm			
Phosphorus (Mehlich III)	23 ppm	50 - 70 ppm			
Calcium (Mehlich III)	4052 ppm	4839 ppm			
Magnesium (Mehlich III)	1247 ppm	415 ppm			
Potassium (Mehlich III)	114 ppm	404 - 809 ppm			
Sodium (Mehlich III)	104 ppm	40 - 119 ppm			
Sulphur (Morgan)	43 ppm	30 - 50 ppm			
Aluminium (Mehlich III)	5 ppm	< 16 ppm			
Silicon (CaCl <sub>2</sub> )	63 ppm	> 100 ppm			
Boron (Hot CaCl <sub>2</sub> )	0.70 ppm	1 - 3 ppm			
Iron (DTPA)	283 ppm	40 - 200 ppm			
Manganese (DTPA)	28 ppm	30 - 100 ppm			
Copper (DTPA)	5.0 ppm	2 - 7 ppm			
Zinc (DTPA)	2.3 ppm	5 - 10 ppm			
Molybdenum (Nitric Acid)	0.24 ppm	0.5 - 2 ppm			
Cobalt (Nitric Acid)	1.80 ppm	2 - 40 ppm			
Selenium (Nitric Acid)	0.30 ppm	0.6 - 2 ppm			
Texture	Clay Loam				
Colour	Red Brown				
<b>BASE SATURATION</b>					
(Levels are not really relevant in soils with a TEC below 5)					
Calcium	58.62 %	70.00 %			
Magnesium	30.07 %	10.00 %			
Potassium	0.85 %	3.00 - 6.00 %			
Sodium	1.31 %	0.50 - 1.50 %			
Aluminium	0.16 %	0.50 %			
Hydrogen	9.00 %	10.00 %			

# Key Mineral Ratios

The **calcium to magnesium** ratio is most important as it determines soil structure and associated gas exchange.

This ratio effectively determines if a soil can **breathe**.

A soil without breath is like an animal nearing death (**oxygen** is required for soil microbes and plant roots while **CO<sub>2</sub>** must exit for photosynthesis).

This is why **calcium** is always the first thing to address when balancing a soil.



# The Key Ratios



The **ideal** calcium to magnesium ratio varies depending on soil density. In a light sandy soil, that ratio should be 3:1 in favour of calcium.

In a heavy, clay soil there is more calcium required to open up that soil, so the ratio should be **7:1** in favour of calcium.

**Calcium** opens (flocculates) the soil as it is a large ion with two positive charges that attaches to tiny, negatively charged particles of clay and pushes them apart. Magnesium is a small ion that does the opposite.



## The Key Ratios

The **magnesium to potassium** ratio is second in importance. You should aim to achieve equal ppm of both elements on a soil test (i.e. 400 ppm K and 400 ppm Mg).

When this happens you have maximum plant availability of both **K and Mg**, but it also stimulates maximum phosphorus uptake.

In **high magnesium** soils, this ratio can never be achieved but there may be more need for applied K and P as a result.



# The Key Ratios



The **phosphorus to sulfur** ratio helps to determine the performance of these two key anions.

Sulfur is often neglected as we no longer receive this mineral **free from the air**.

Sulfur is critical for healthy root growth, **protein formation** (essential for plant immunity), and chlorophyll density.

When we lift sulfur levels to match P levels, there is a substantial improvement in **soil and plant health**.



## The Key Ratios

The **phosphorus to zinc** ratio is next in importance. Here we like to see 10 parts of P for every 1 part of Zn.

If this this ratio is achieved there will be **maximum performance** of both minerals.

Too much of either **inhibits** the other.

Don't ever try to achieve this ratio in **high phosphorus** soils as the zinc will tie up other elements.





# The Key Ratios



The **potassium to sodium** ratio is critical to ensure availability of the second most abundant mineral in the plant (K).

If sodium **exceeds potassium** in terms of base saturation, then sodium becomes a destructive influence.

This can be a common problem where **recycled, saline water** is involved.

# The Key Ratios

The **iron to manganese** ratio is important to ensure optimum uptake of both minerals.

Ideally, iron should always be **higher than manganese** on a soil test.

The ratio should be between **1.1:1 and 2:1** in favour of iron.

Unfortunately, the uptake of both minerals is seriously impacted by **glyphosate**.



# Soil Mineral Management

## Understanding the Terminology

- **Cation exchange capacity (CEC)** – offers an immediate guideline as to storage capacity in your soil.

### Key guidelines:

1. Ignore base saturation percentages in light soils with a **CEC below 5**. Here, the goal is **500 ppm** of calcium, and **120 ppm** of both magnesium and potassium.
2. The term '**cation exchange**' refers to an exchange that takes place when a cation is removed from the clay colloid by the hungry plant. The plant must maintain electrical balance so it 'exchanges' the cation **hydrogen** with whatever cation it strips from the colloid. Hydrogen is the **acid** element.



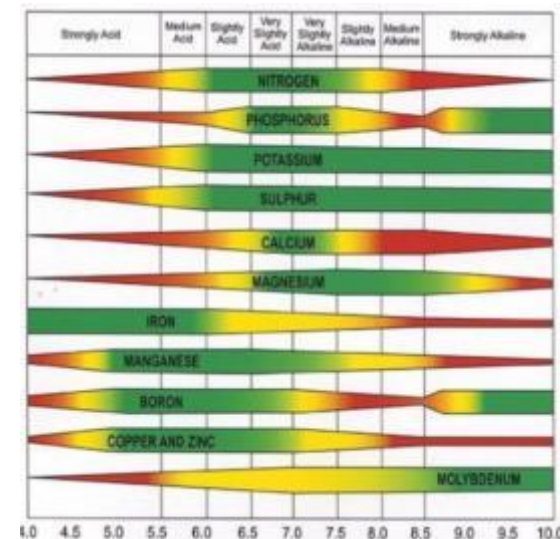
Soil Material	Storage
Sand	2 – 3 CEC
Silt	5 – 7 CEC
Heavy Clay	30 – 60 CEC
Humus	250 CEC
<u>Humic Acid</u>	450 CEC
<u>Fulvic Acid</u>	1400 CEC



# Soil Mineral Management

## Understanding the Terminology

3. **TEC** – refers to total exchange capacity. It means that the percentage of the non-nutrient mineral hydrogen has been factored into the equation.
  - Without the inclusion of **hydrogen** in the mix, you can be misled into thinking you have an adequate cation balance, when in actual fact, half of your clay storage may be filled with this acid-forming, non-nutrient mineral.
4. **Soil pH** – minerals are most available at a soil pH of 6.4
  - It is much easier to correct an acid soil than to address a **highly alkaline** soil.
  - Often, you will need to bypass the soil and **foliar spray** minerals like iron, manganese and boron when the soil pH is above 8.
  - **Manganese** and **boron** are critical before flowering. Manganese impacts seed production while boron affects multiple aspects of reproduction.



# Soil Mineral Management

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## Understanding the Terminology

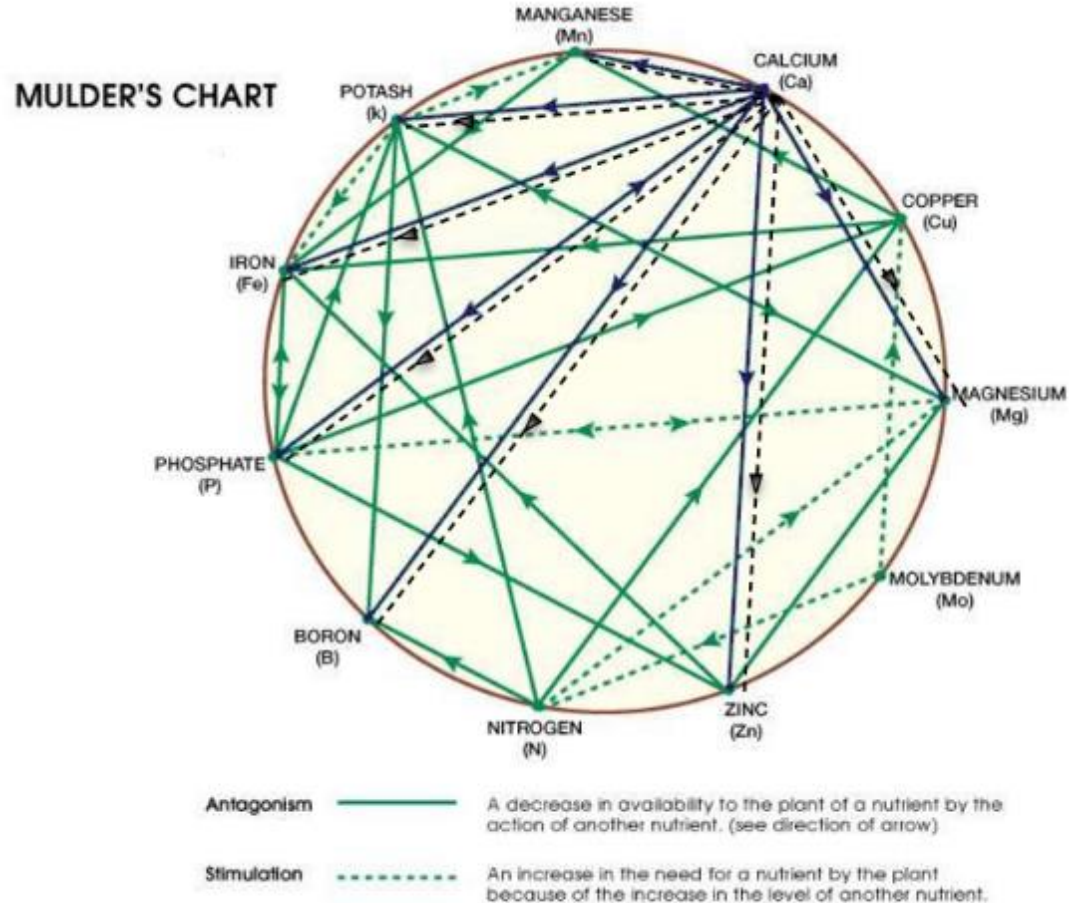
5. **Organic matter** – This is the most important guideline for high production fertility.
  - **Soil structure**, improved water and nutrient retention, crop resilience, soil stability and enhanced infiltration are some of the many **benefits** of organic matter.
6. **Estimated nitrogen release** – refers to the fact that organic matter stores both ammonium and nitrate nitrogen, and releases a percentage of its N storehouse during the crop cycle.



## Understanding the Key Minerals

7. **Calcium** – This is the mineral that should be addressed before all else.

- We refer to calcium as “**the trucker of all minerals**”, as it directly impacts, both negatively and positively the uptake of **seven** different minerals
- Calcium is also the doorman that determines mineral availability at the **cell membrane**.
- It is the **most important** mineral in humans, animals, the soil, and your crop.





## Understanding the Key Minerals



### 7. Calcium –

- In the soil calcium is critical for **gas exchange**.
- In the plant, it determines **cell strength** and associated resilience.
- It is also a key player in **crop quality, shelf life, and the availability of nutrients**.
- It also impacts the activity of beneficial microbes, via increased oxygen through **flocculation**.



# Soil Mineral Management

## Understanding the Key Minerals

### 7. Calcium –

- Minerals vary in their **mobility**. N, P & K are the most mobile. Calcium is the least mobile of all minerals.
- That means it **moves very slowly** into the plant, around the plant, and into the fruit and seeds.
- Many mineral **excesses**, including nitrogen, magnesium and potassium, inhibit calcium milk intake.
- **Foliar fertilising** is essential for good calcium supply.
- Calcium is one of the critically important minerals we call “**The Big Four**”.

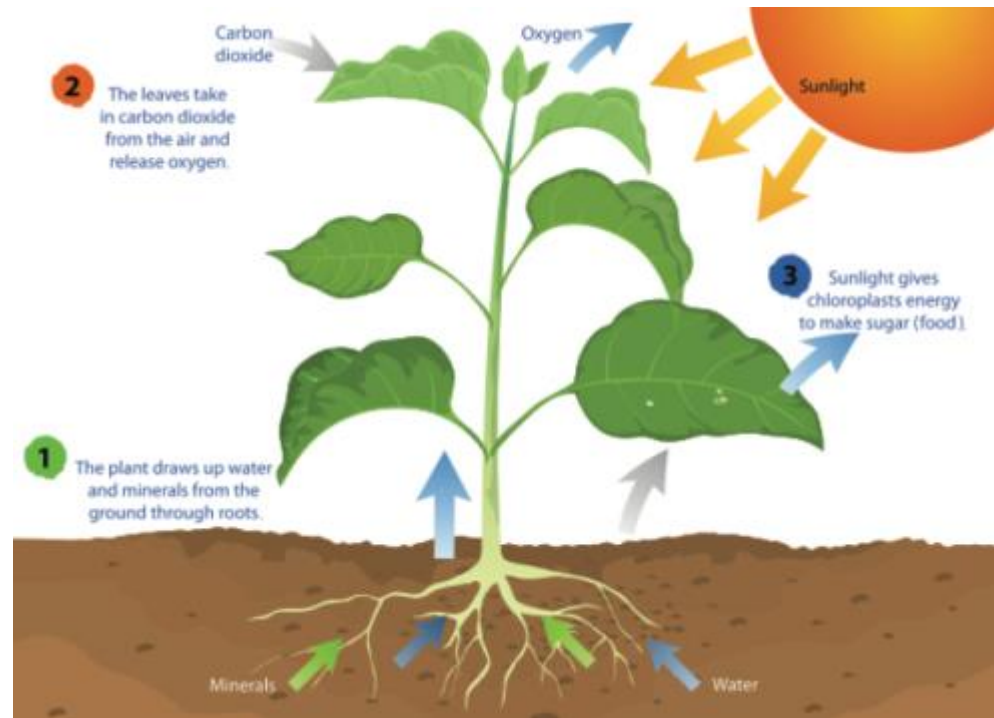


# Soil Mineral Management

## Leaf Testing – Removing the Guess Work

### Understanding “The Big Four”

- When **four minerals** are maintained at luxury levels, according to a leaf test, there can be a very productive outcome.
- The **big four** minerals include; calcium, magnesium, phosphorus and boron.
- All four are directly related to the most important of all plant processes, **photosynthesis**.
- The sad story is that in **30%** of all of the leaf tests we analyse, all four of these important minerals are deficient (let alone at their optimum level).



# PLANT THERAPY™

**DATE:** 02.04.2009

**LAND USE:** Rye Grass

**NAME:** Mr Sample

**Paddock:** Sample Paddock

**ADDRESS:** -

**SAMPLE REC:** 28.03.2009

**CONTACT NO:** XX XXXX XXXX

ELEMENT OR CATEGORY	YOUR LEVEL	ACCEPTABLE RANGE	DEFICIENT	ACCEPTABLE	EXCESSIVE OR TOXIC
N - Nitrogen	4.5 %	2.3 - 4.2 %			
P - Phosphorus	0.15 %	0.2 - 0.5 %			
K - Potassium	3 %	1.5 - 3.5 %			
S - Sulphur	0.54 %	0.2 - 0.45 %			
Ca - Calcium	0.11 %	0.2 - 0.6 %			
Mg - Magnesium	0.14 %	0.16 - 0.4 %			
Na - Sodium	0.3 %	< 0.7 %			
Cu - Copper	8 ppm	5 - 12 ppm			
Zn - Zinc	20 ppm	15 - 50 ppm			
Mn - Manganese	100 ppm	50 - 300 ppm			
Fe - Iron	55 ppm	50 - 60 ppm			
B - Boron	3 ppm	5 - 15 ppm			
Mo - Molybdenum	0.16 ppm	0.15 - 0.5 ppm			
C - Carbon	%	N/A	N/A		

# Soil Mineral Management

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## Understanding the Key Minerals

**8. Phosphorus** – we find Mehlich III extraction to give the best overall guideline. We aim for 50-70 ppm.

### Guidelines:

- Just **27%** of applied DAP/MAP is utilised. The balance becomes part of a massive frozen bank account.
- There is little sense in applying water soluble phosphate to **pasture** and **orchard crops**, considering the loss factor.
- In field crops, DAP/MAP should always be stabilised with **Soluble Humate Granules (SHG)**. This practice also buffers the burning impact on mycorrhizal fungi and plant roots. **5%** of SHG does the job while also providing other benefits.

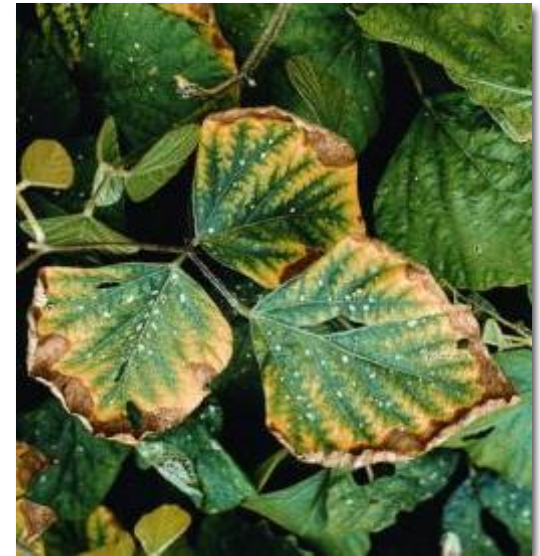




## P Deficiency in Plants

- Stunted shoot & root system
- Leaves lack lustre
- Yellowing
- Blue green or purplish leaves
- Symptoms appear first in older leaves

P is very mobile in the plant



# Refractometer = total dissolved solids

- A good guideline to P levels as sugars can not be produced without phosphate.
- The higher the brix the stronger the plant, good energy reserves to fuel any demand response (e.g. pest attack, onset of flowering or a growth spurt on a sunny day after rain).
- Taste, Nutrition, Colour, Size, Shelf Life.



# Brix to Control Insects

## INCREASED PEST & DISEASE RESISTANCE

- P provides fuel for energy expensive plant defence systems.
- This includes **anti-fungal agents** and production of **biochemicals** which are distasteful to insects.
- Phosphate is essential for sugar production and **increased sugar levels** are **unattractive** to insects. i.e. it can convert to poisonous **alcohols** in their bodies.



# Ravaging the Root Zone

Study the roots to evaluate the biological **impact** of harsh fertilisers.

- Look for fuzzy coating around the roots and how it holds onto the soil.
- This **fuzzy coating** represents billions of fungi and bacteria and their sticky exudates and hyphae effectively hold the soil to the roots.





# Combining Soluble & Slow Release

DAP/MAP with humates plus Guano Granules or a similar **Soft Rock** phosphate alternative.



### RATIONALE:

- The acid phosphate is buffered and stabilised by the humates.
- The guano offers citrate soluble and slow release phosphate so there is a complete release pattern throughout the crop cycle.
- **Soil life** is both protected and stimulated with this fusion approach.

# Soil Mineral Management

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## Understanding the Key Minerals

### 7. Phosphorus

- Strategies to reclaim locked up phosphorus include: **compost**, legumes, fulvic acid, mycorrhizal fungi, and **stubble digestion** programs.
- Magnesium can stimulate the uptake of phosphorus, while high **potassium** can limit P availability.

### 8. Magnesium – this mineral is the centre of chlorophyll, so a shortage will mean substandard photosynthesis.

- High magnesium creates tight closed soils where **gypsum** is so important.
- **Magnesium sulphate** should never be used in the soil. It is only applicable as a foliar spray.



# Soil Mineral Management

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## Understanding the Key Minerals

9. **Potassium** – this super mobile mineral does not become part of cell structure, but races around the plant to where it is required.
- K is involved in stomatal opening, stem strength, vegetative growth and, most importantly, the **sizing up** of grains and fruit.
  - The ‘**money mineral**’ is best managed by monitoring top and bottom leaves, and ensuring that there is no more than **10% variation** between the two sites.
  - **Potassium sulphate** is always the best option even though it costs more.
  - A **potassium metre** is the best tool to manage this mineral.

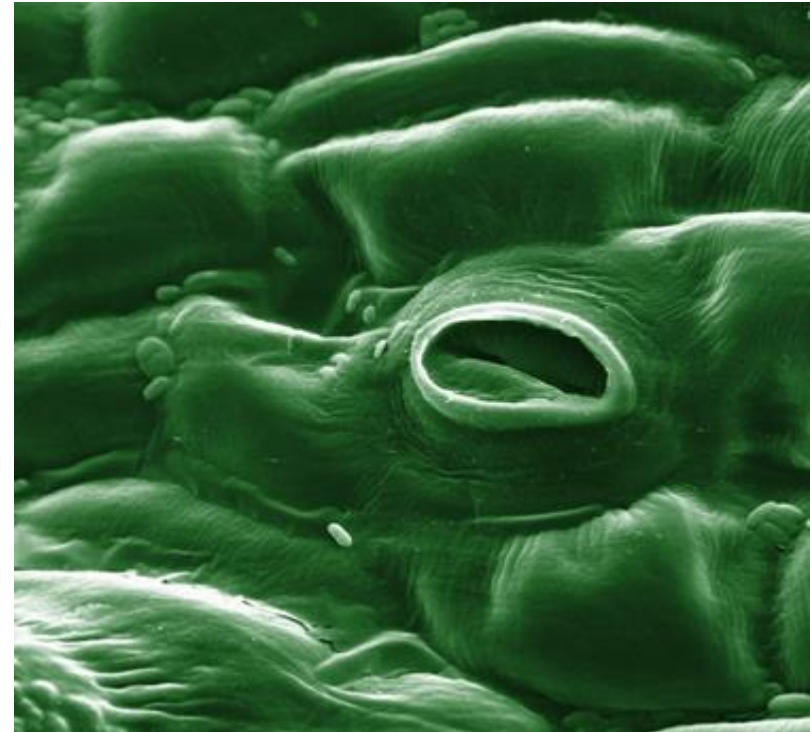


# Soil Mineral Management

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## Understanding the Key Minerals

10. **Sulphur** – We used to get our sulphur for free, until acid rain damaged forests and waterways, and sulphur emissions were banned.
- **Organic matter** is the only thing that stores highly leachable sulphur in the soil, and we have lost 2/3 of our humus.
  - Sulphur is about **protein formation**. Protein drives plant, animal, and human immunity.
  - Sulphur availability is impacted by other **major anions** (vice versa). The 1:1 sulphur to phosphorus ratio is critical.
  - **Elemental sulphur** can give a more profound response because the gas emitted while it is biologically converted to sulphates can be absorbed through the stomates.





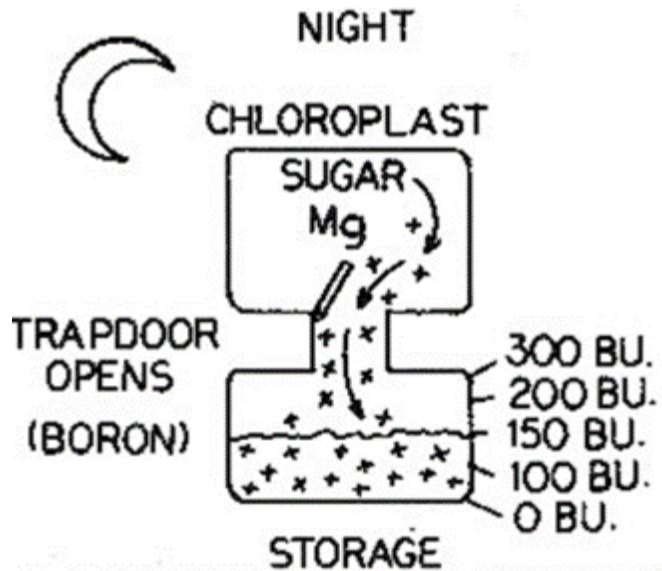
# Soil Mineral Management

## Understanding the Key Minerals

11. **Silica** – This mineral strengthens the cell wall to resist insects and disease.
- The plant available form is **missing** in most soils, and appears to be linked to the chemical extractive model (**100ppm** is needed).
  - Silica also prevents **lodging**, improves nutrient translocation, combats mineral excesses (sodium), improves photosynthesis, and increases resistance to both biotic and abiotic **stress** (heat, cold, drought, transplant).
  - Most importantly, silica is a major **immune elicitor**, hence the link to increase yields when we address the shortage.
  - Liquid, micronised **diatomaceous earth** is the most versatile silica source.



# Soil Mineral Management



## Understanding the Key Minerals

12. **Boron** – This is the most important calcium synergist. It is a highly leachable anion, and is only stored on the humus colloid. Ideal soil levels are 1 ppm – 3 ppm.

- Boron is deficient in most leaf tests we analyse, and it is most effective as a **foliar spray** directly **before flowering**. It is critical for grain and fruit formation.
- Boron opens the **trap door** in the chloroplast to allow translocation of glucose around the plant and down to the roots. 30% of this is fed to soil microbes. If the door remains shut, the microbes suffer.
- A **refractometer** can help identify a boron deficiency (Brix should always be lower in the morning).
- Boron should always be applied with **humic acid** to stabilise and magnify the mineral.



# Soil Mineral Management

## Understanding the Key Minerals

13. **Zinc** – 5-10 ppm is required. This mineral governs leaf size, so a deficiency will always be costly.
- The **phosphate to zinc ratio (10:1)** is critical for availability, and for the energy roles of both minerals.
  - Zinc sulphate can be easily chelated with **fulvic acid** to create a zinc fulvate.
  - **Copper fungicides** can induce zinc deficiencies.
14. **Copper** - Is about protection from fungal disease, chlorophyll density, stem strength, and protein formation.
- **Copper fungicides** are the least sustainable method of disease control. It is far better to get copper **into the plant** efficiently rather than drenching soil and plant with a heavy metal (Copper Shuttle™).



Zinc Deficiency in Wheat



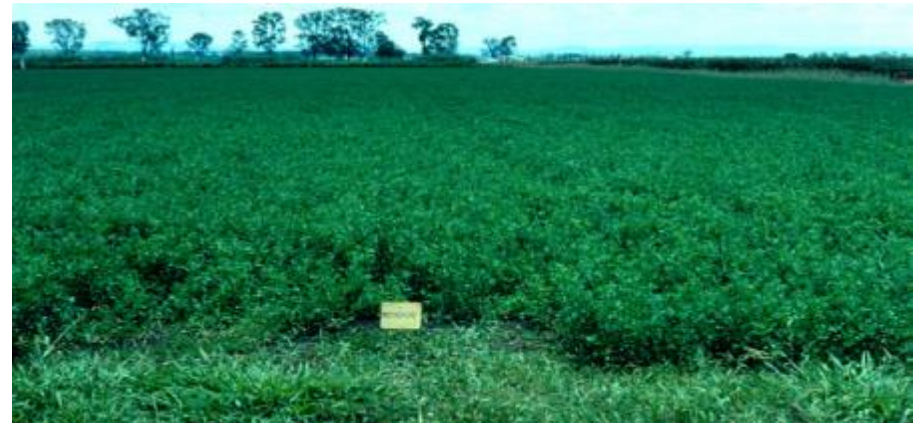
# Soil Mineral Management

## Understanding the Key Minerals

14. Molybdenum – This mineral is needed to fix nitrogen and to convert nitrates to proteins. Three molybdenum-based enzymes are linked to stress resistance.
- 80% of the soils we test are molybdenum deficient. We need minimum levels of 0.5ppm in the soil.
  - It is a highly leachable anion.
  - Molybdenum deficiency is widespread on **legumes** and **maize** in acid soils. This is related to the antagonistic effect of iron.
  - Light, **sandy**, low humus soils are often molybdenum deficient.

Potential crop improvements in Waitaki County, New Zealand, by the use of molybdenum fertilisers. From LOBB [170]

CROP AND SOIL	CONTROL, NO MOLYBDENUM	YIELDS WITH MOLYBDENUM	% INCREASE
Lucerne on sandstone soil	3,060	13,920	355
Lucerne on clay soil	13,664	19,488	43
Rape on sandstone soil	7,616	53,536	603
Rape on clay soil	8,288	14,784	78
<u>Pasture on clay soil:</u>			
a	3,300	6,233	89
b	2,038	9,357	359
c	3,561	17,450	390
d	995	1,505	51
e	4,968	7,028	44
f	2,080	2,868	38
g	13,700	20,500	50
h	12,200	18,800	54





A herd of black cows is grazing in a lush green field. The cows are in the foreground and middle ground, with some looking towards the camera. The background shows a line of trees and a cloudy sky. The text is overlaid on the lower half of the image.

## **In Conclusion...**

**Understanding the interplay between minerals is essential to maximise productivity.**

**Soil tests are important, to avoid driving blind, but you must understand the data.**

**Knowledge is more important than inputs in the regenerative approach.**